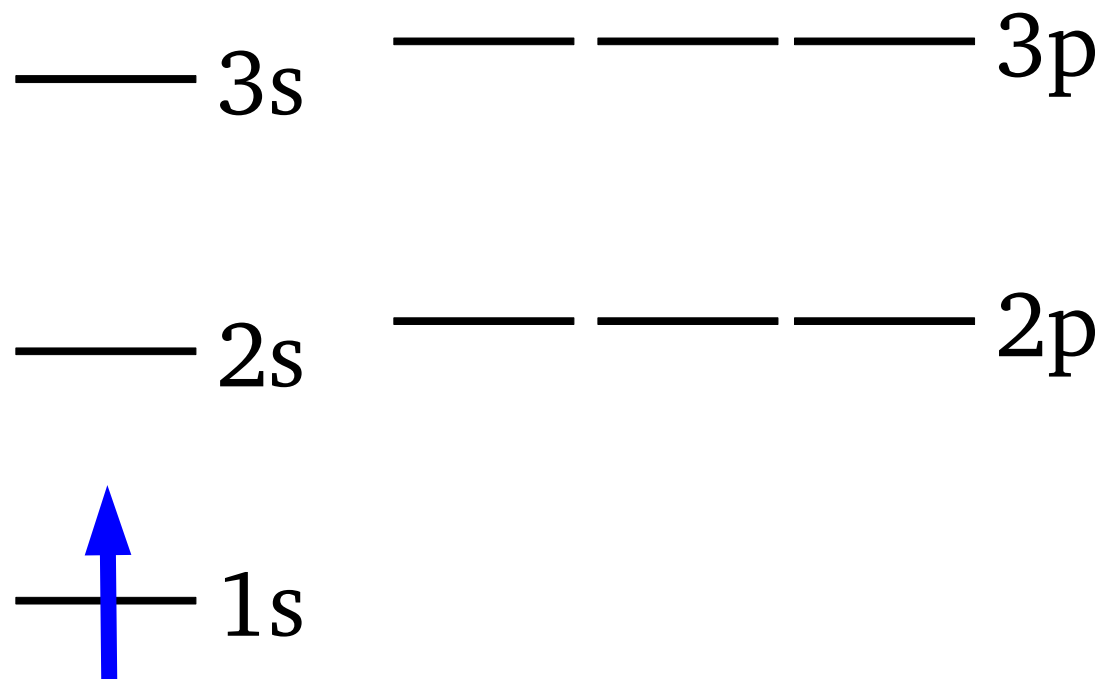


# Configuration électronique des 18 premiers éléments de la classification périodique

correction

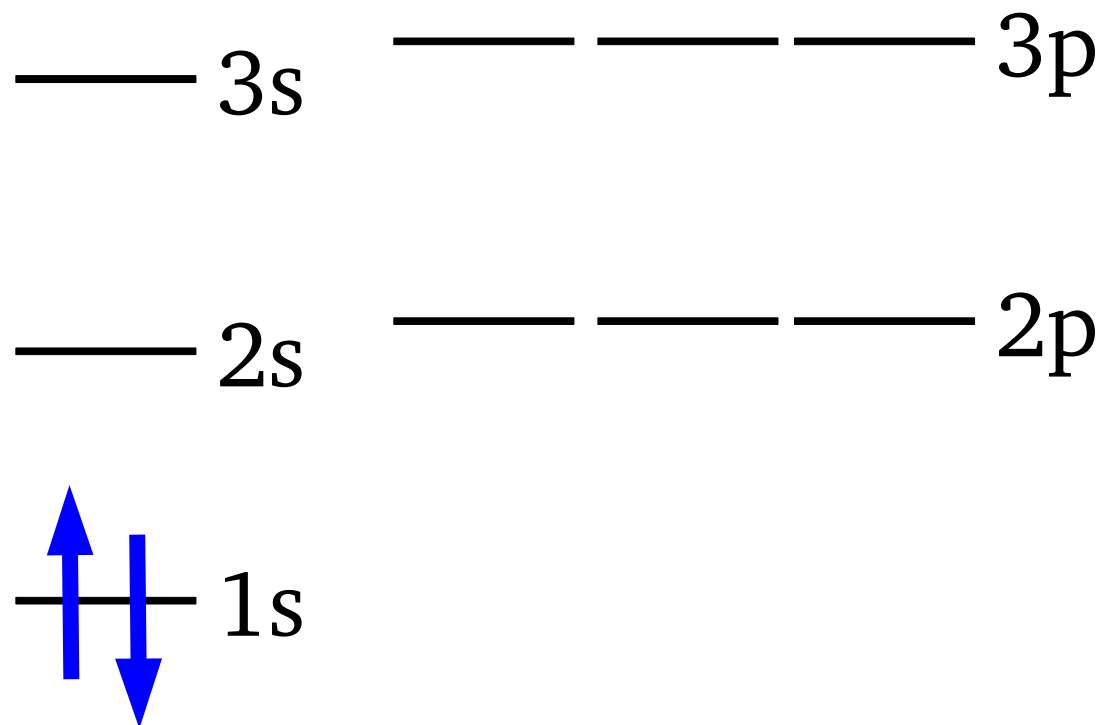
# 1 *hydrogène*

H



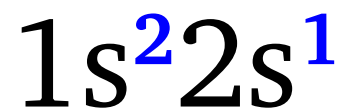
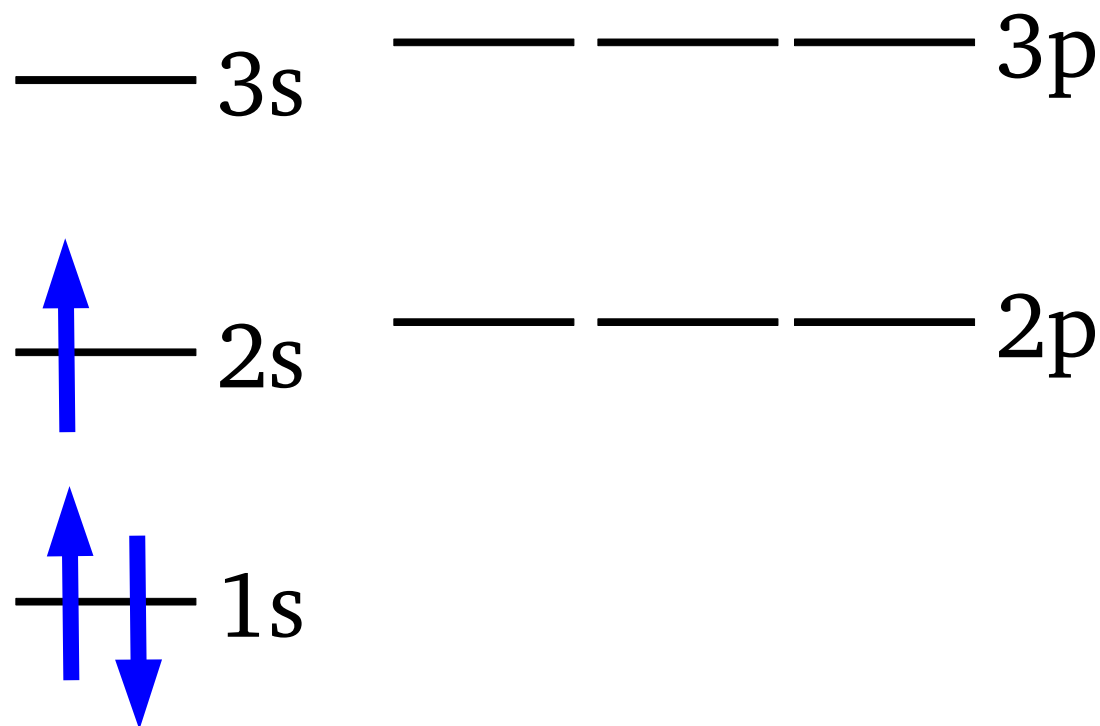
# 2 *hélium*

He



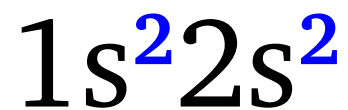
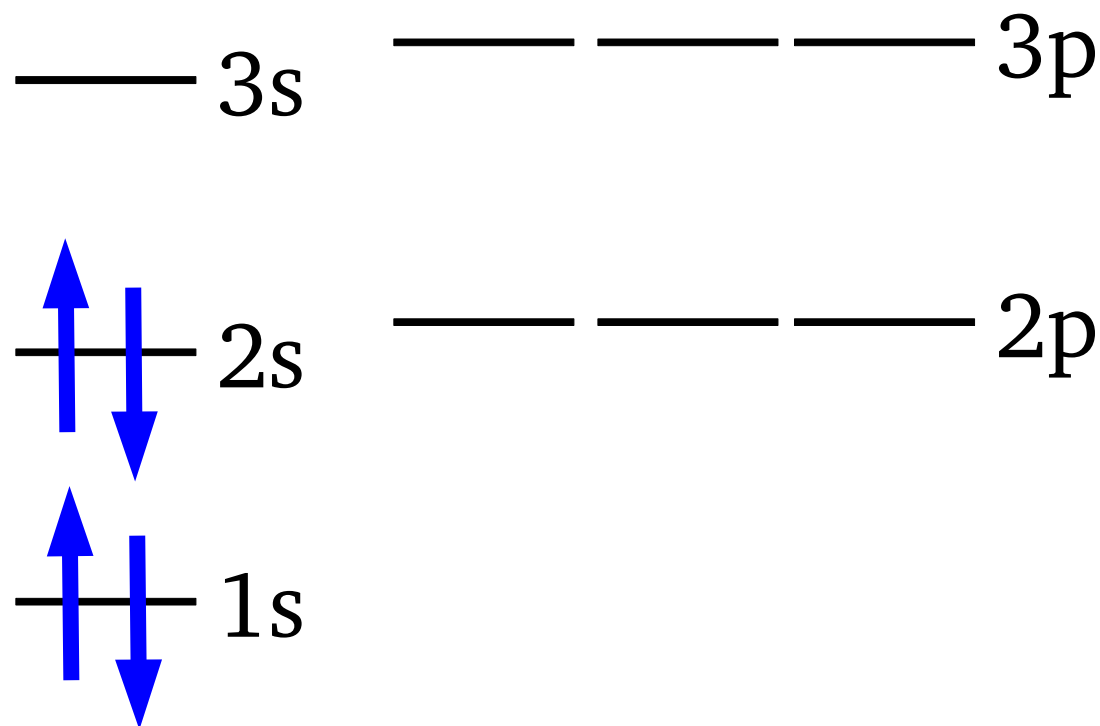
**3** *lithium*

**Li**



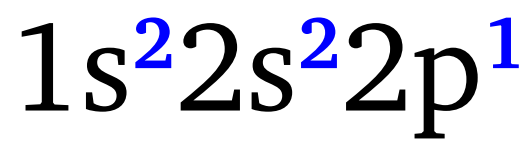
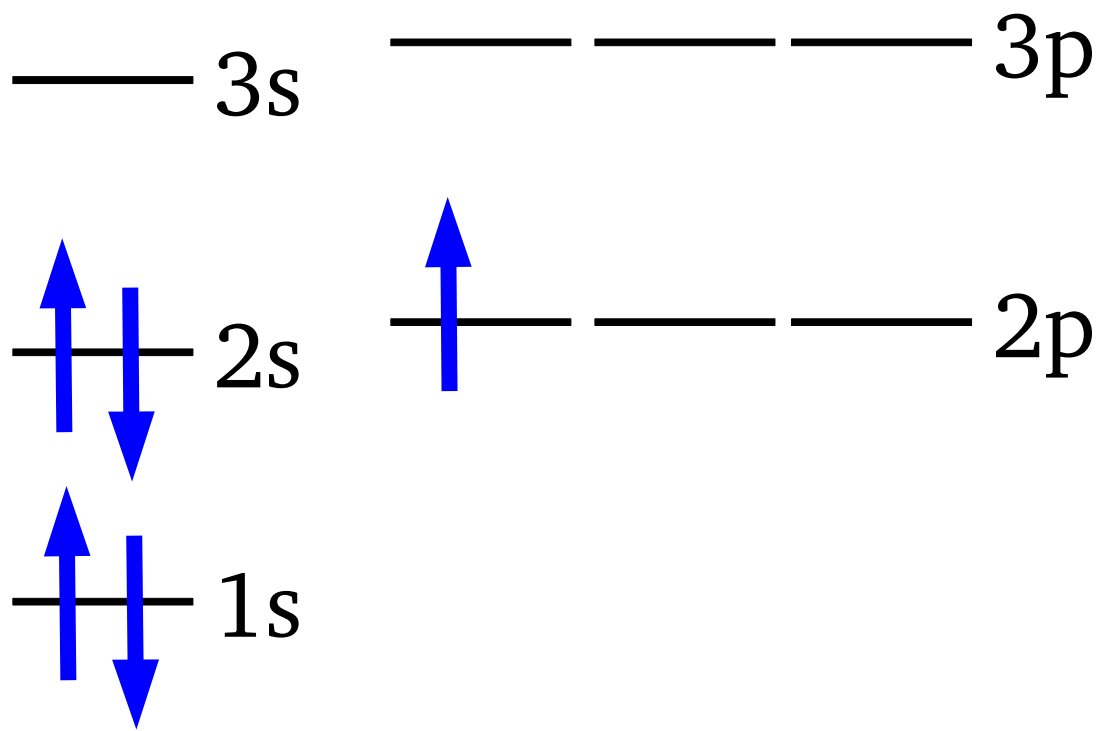
4 *béryllium*

Be



**5** *bore*

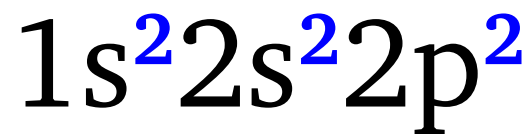
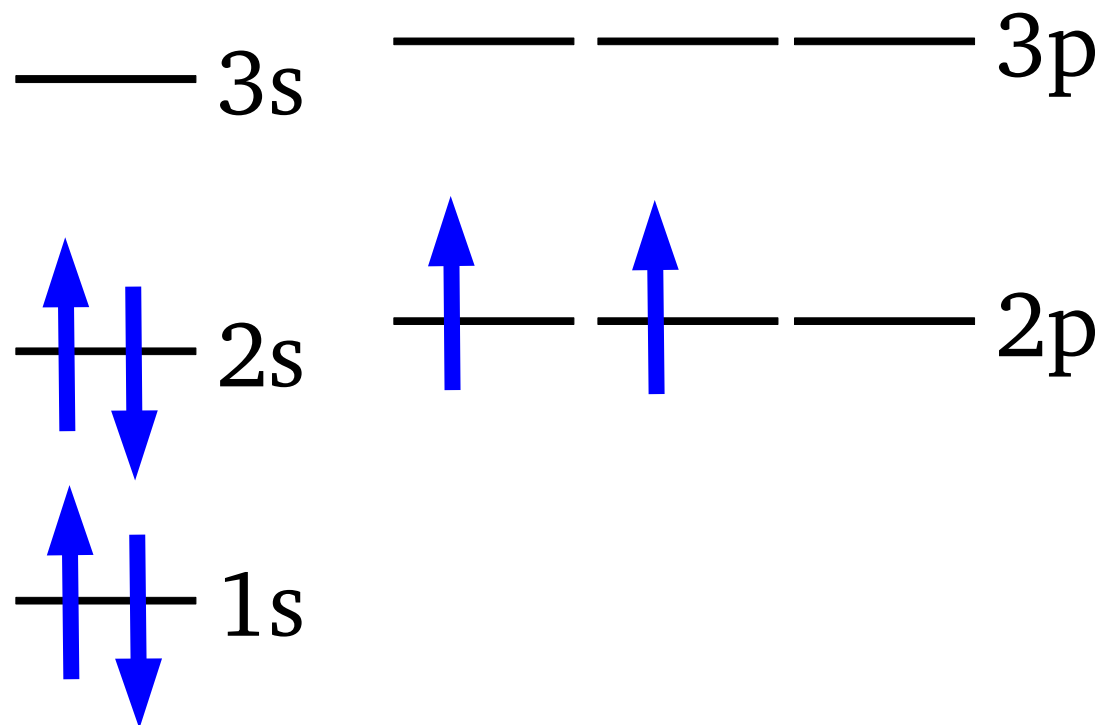
**B**



6

*carbone*

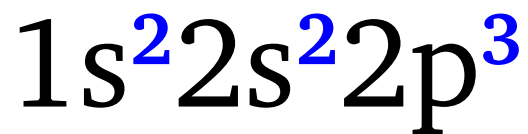
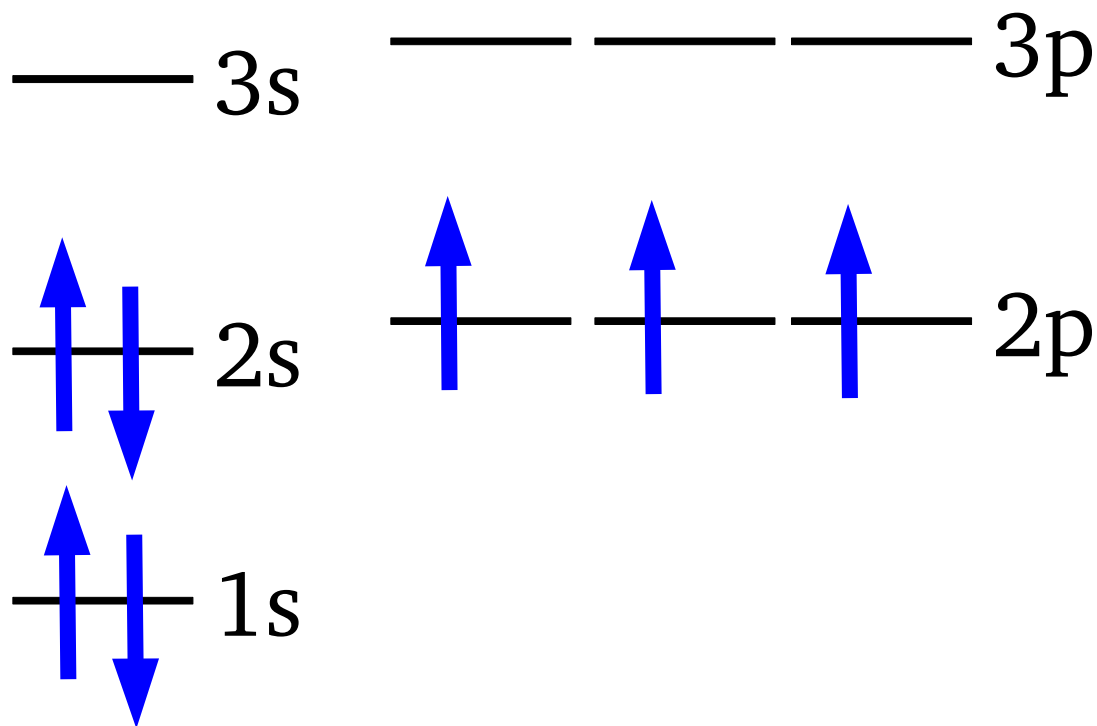
C



7

*azote*

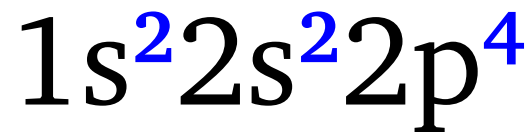
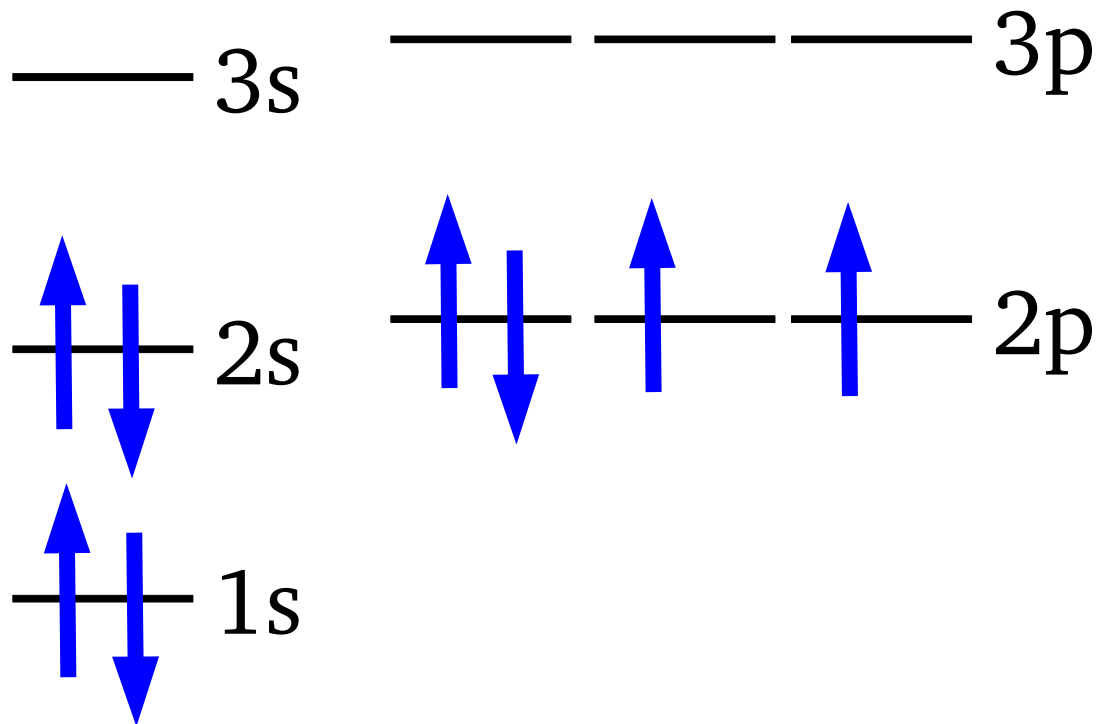
N



8

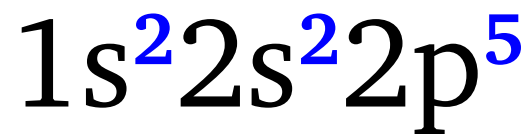
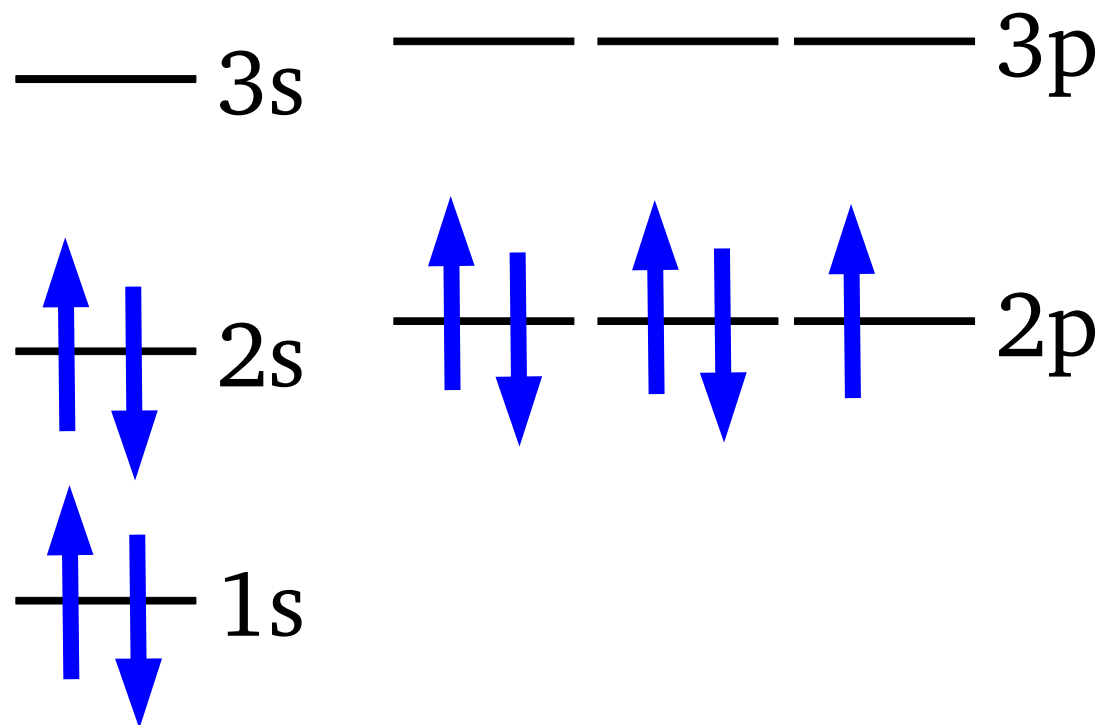
*oxygène*

O



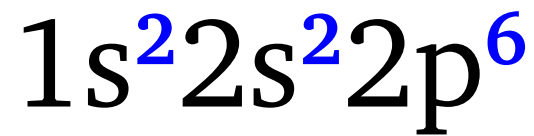
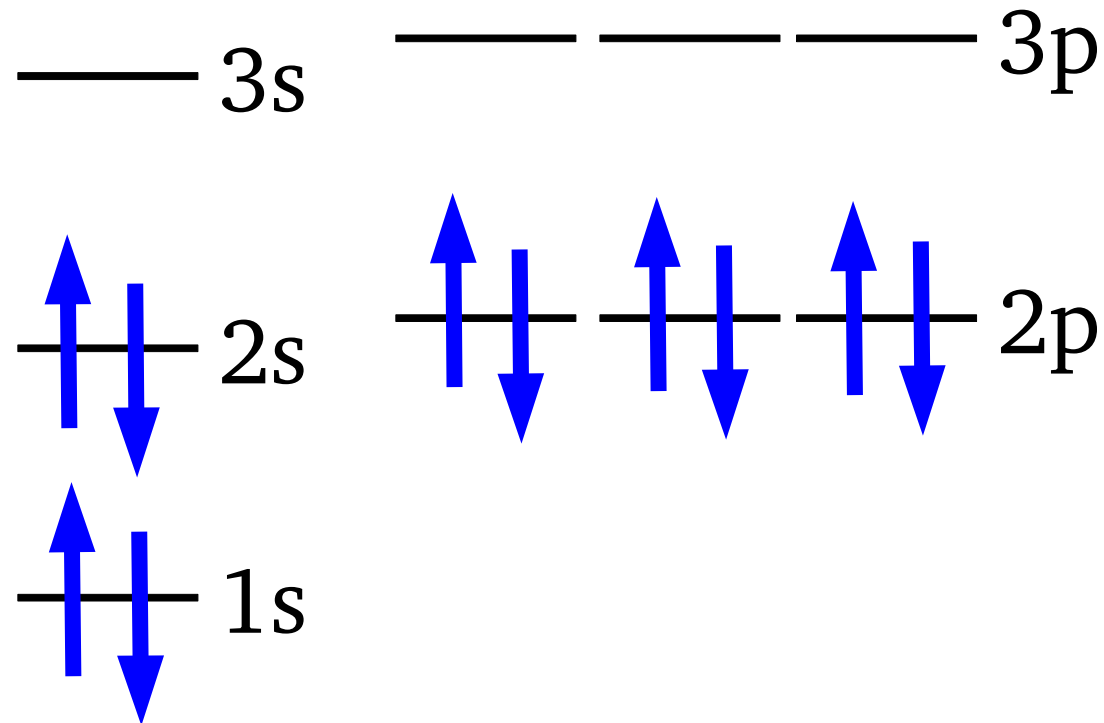
9 *fluor*

**F**



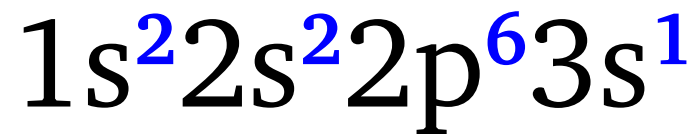
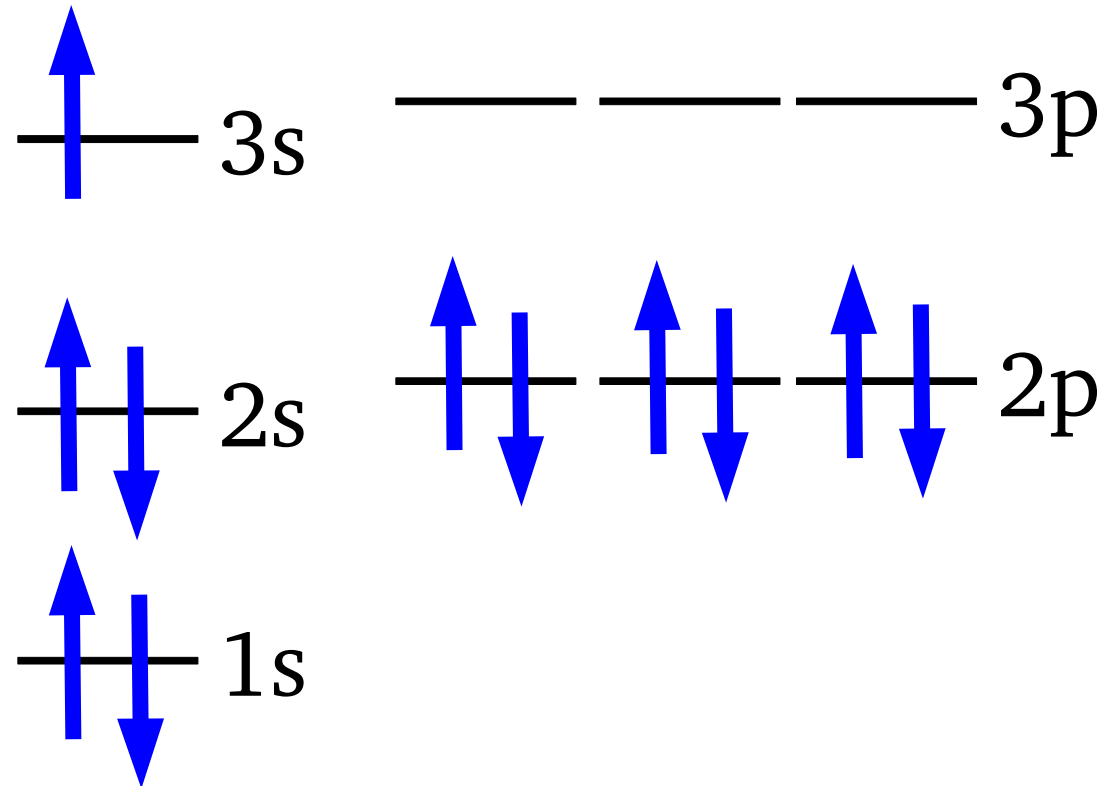
10 *néon*

Ne



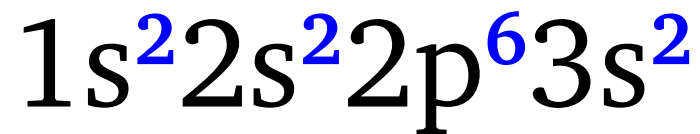
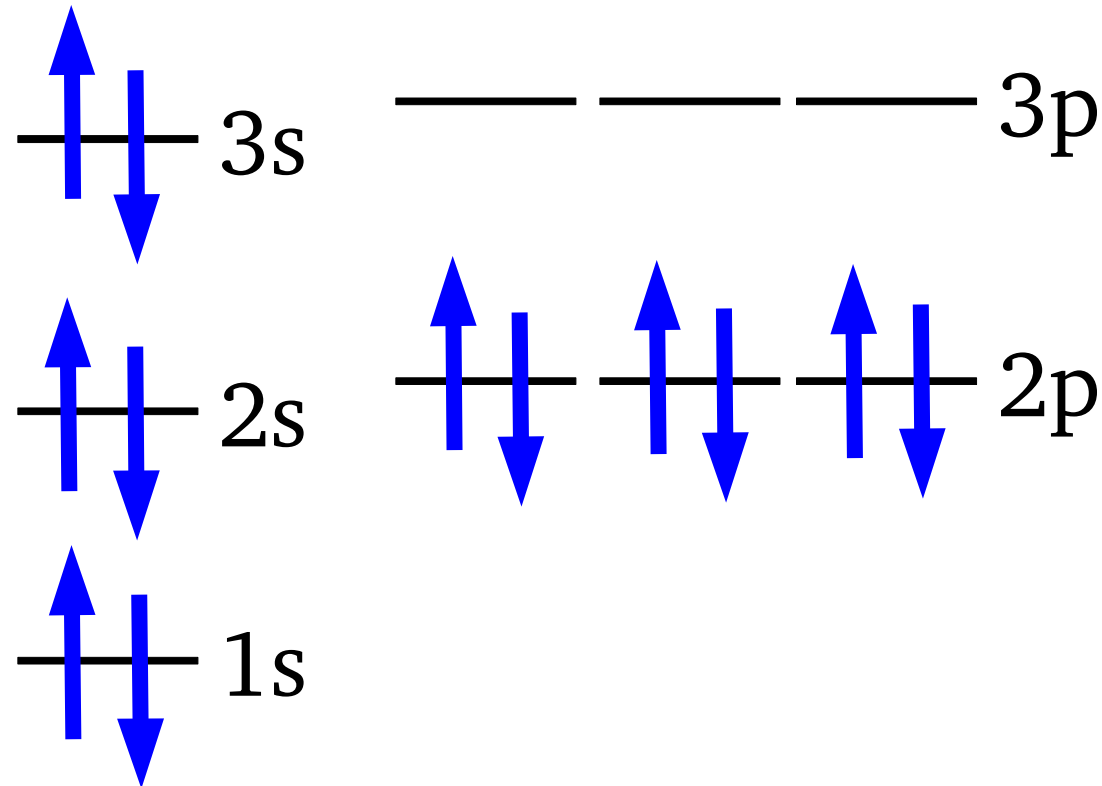
**11** *sodium*

**Na**



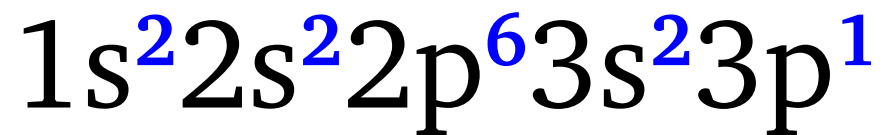
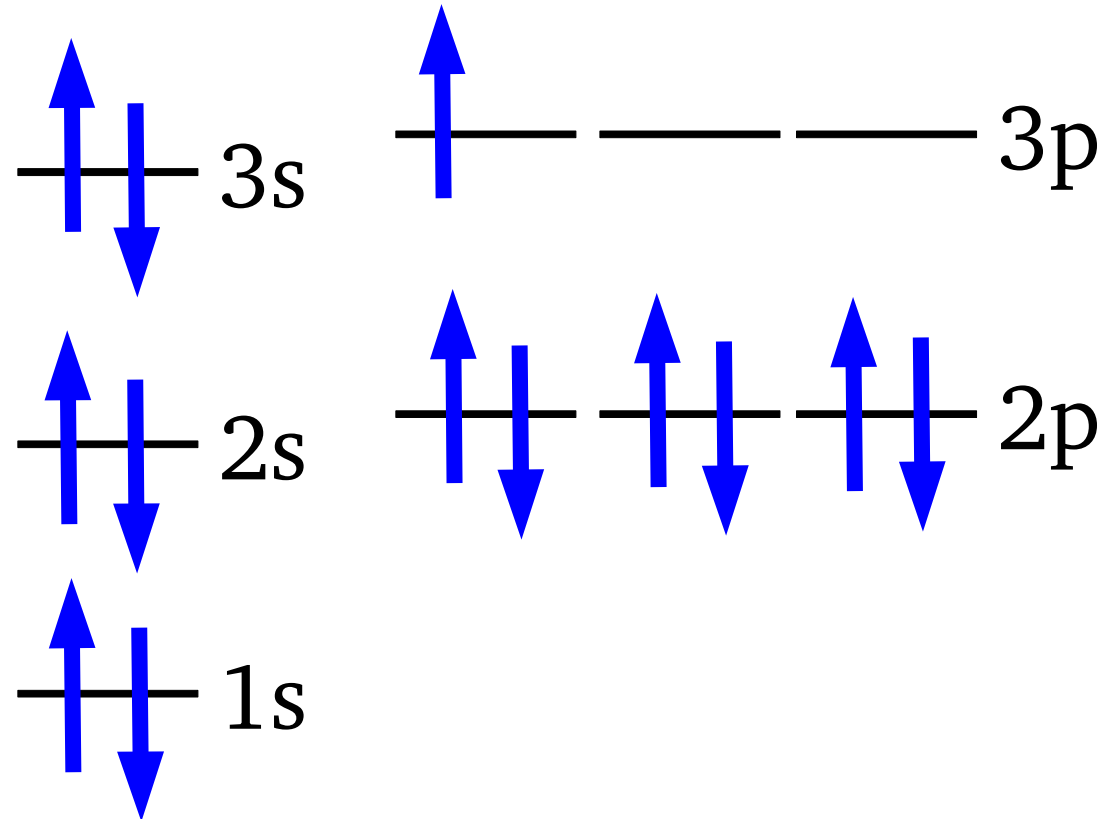
**12** *magnésium*

**Mg**



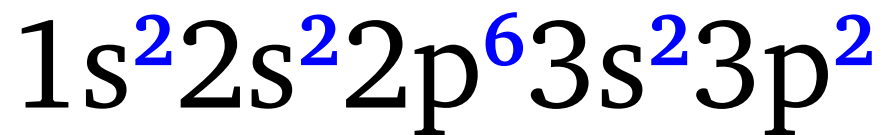
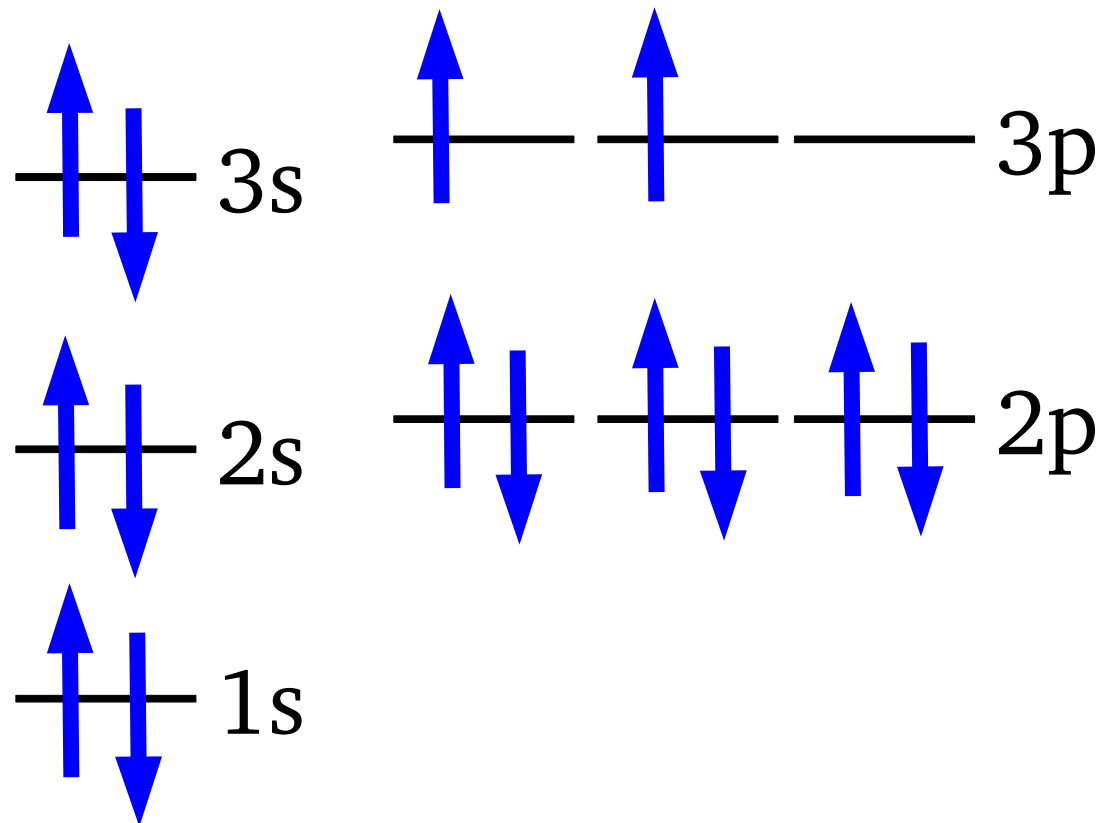
**13** *aluminium*

**Al**



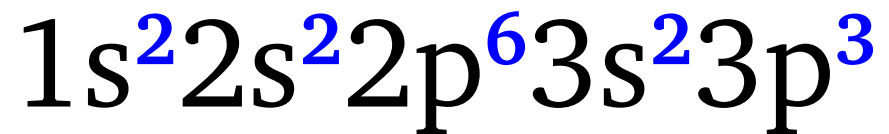
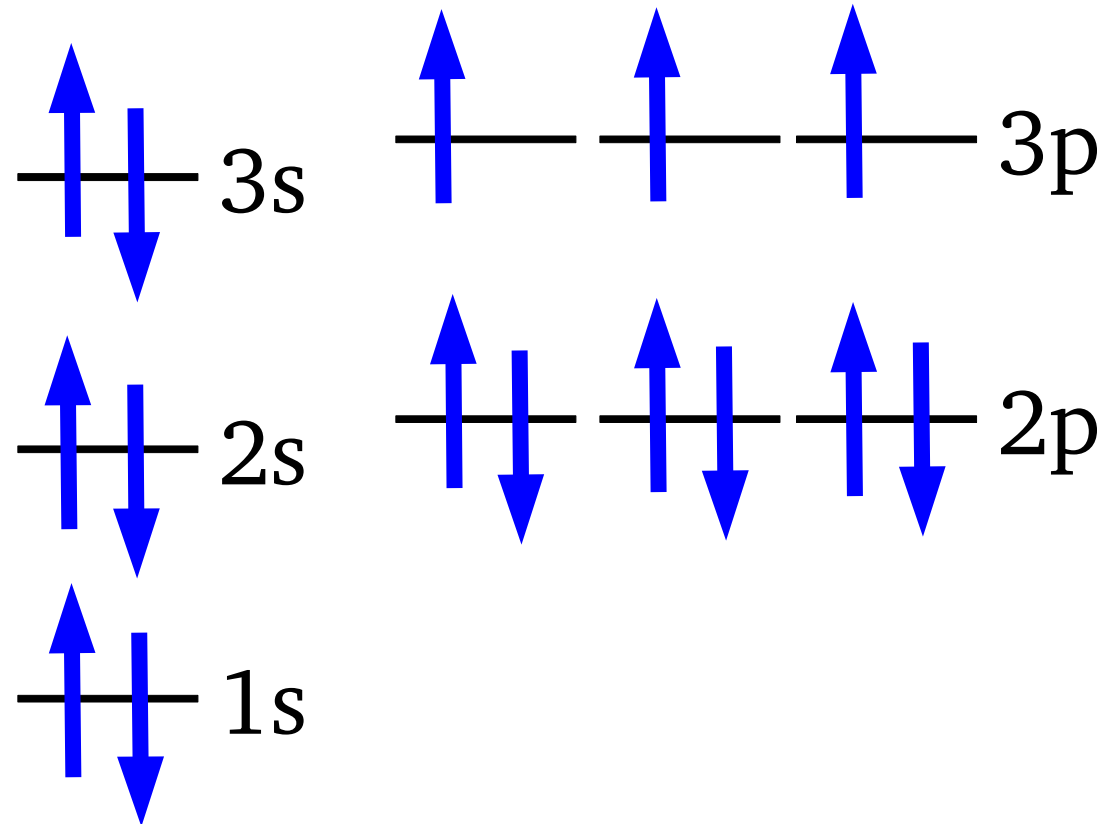
14 *silicium*

Si



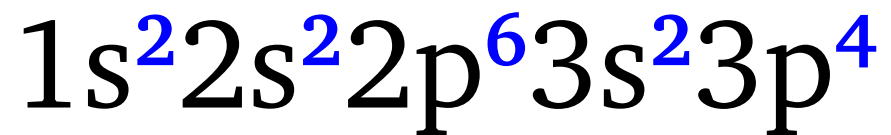
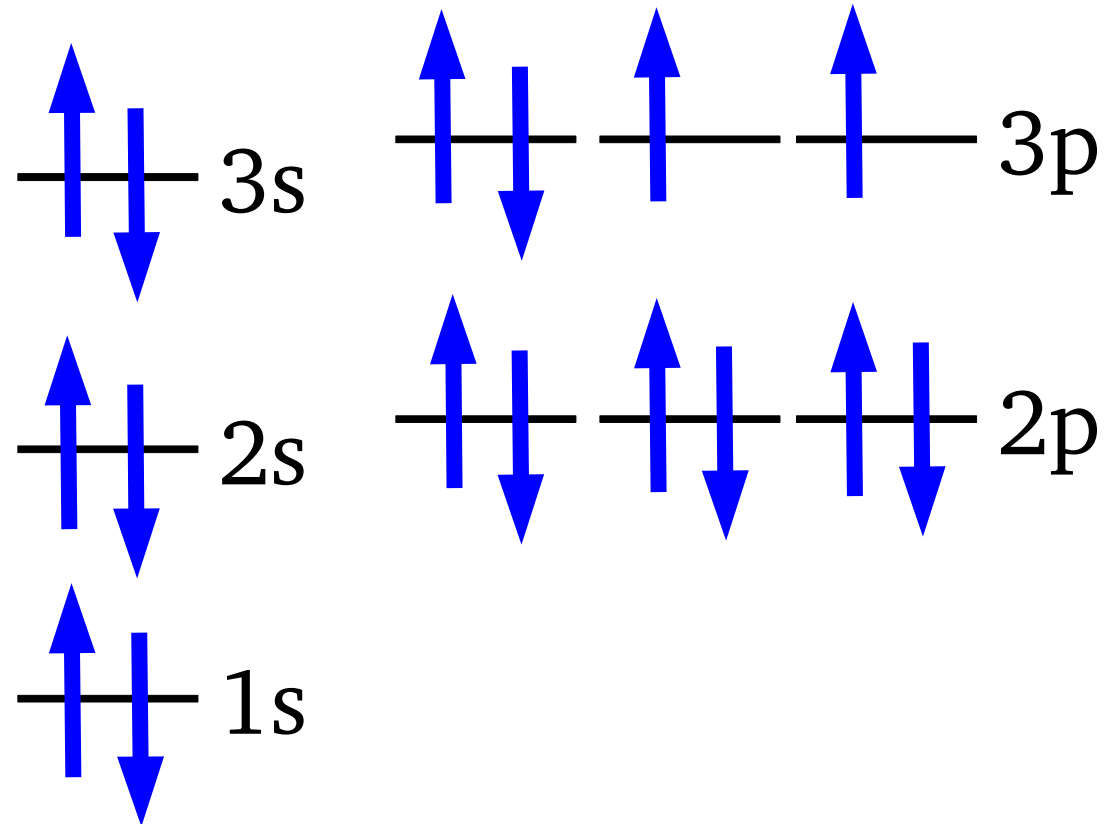
15 *phosphore*

P



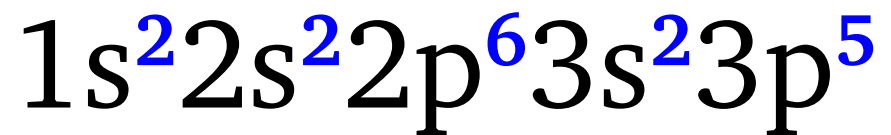
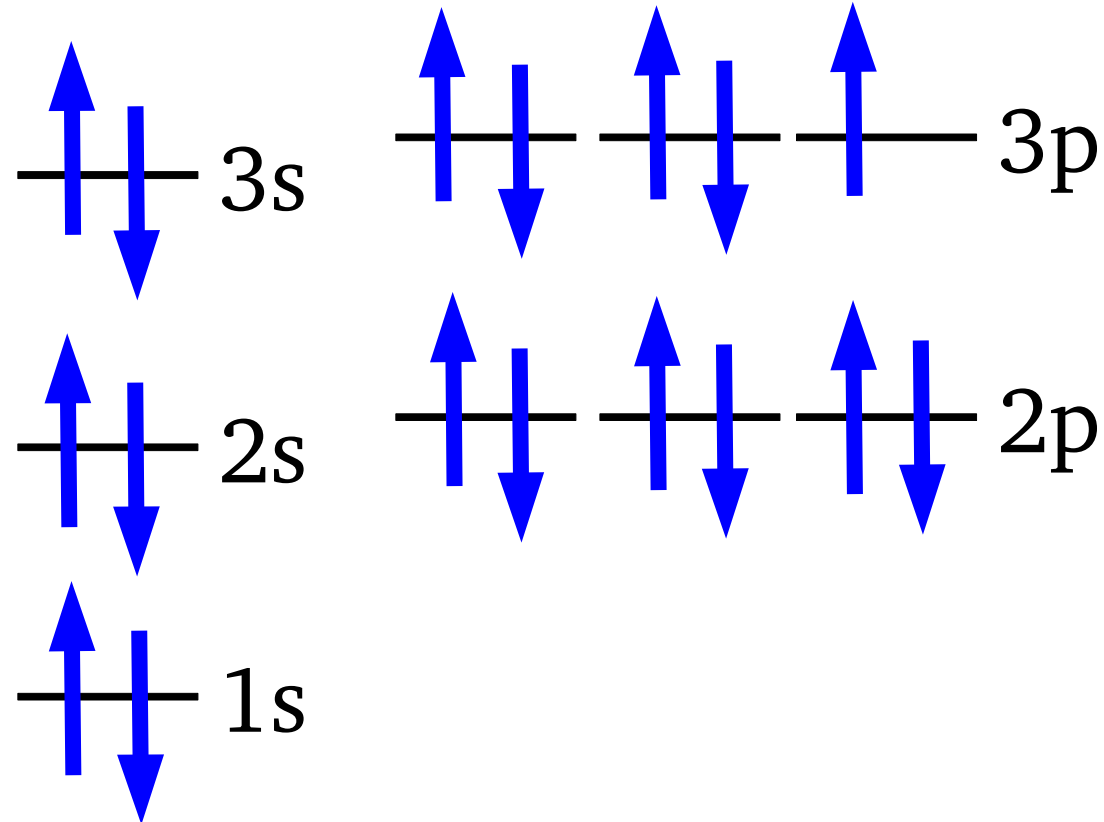
16 *soufre*

S



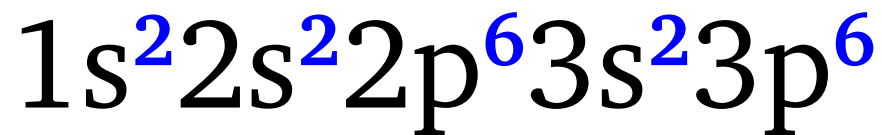
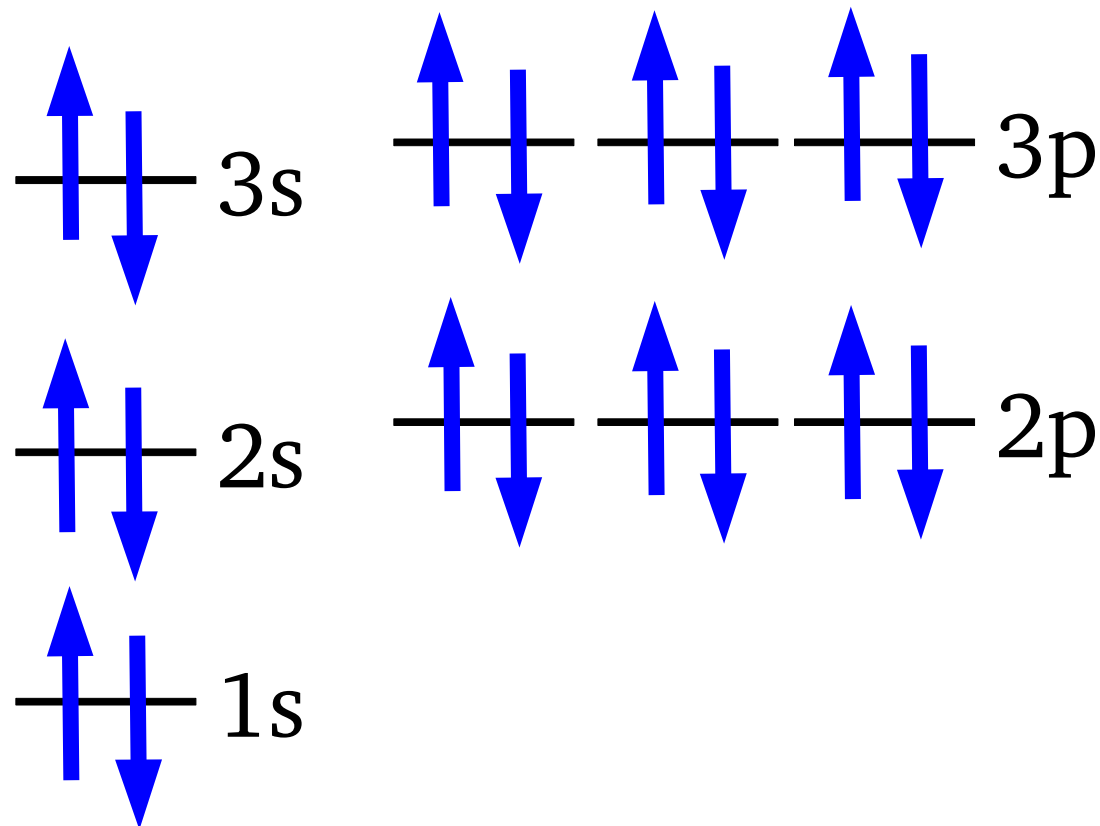
17 *chllore*

Cl



18 *argon*

**Ar**

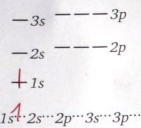


*découpez les 18 cases*

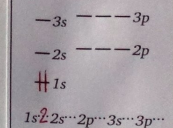
**puis**

*collez-les sur une feuille A3 en  
reformant l'aspect du tableau de la  
classification périodique des éléments ...*

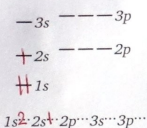
1 hydrogène H



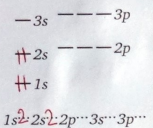
2 hélium He



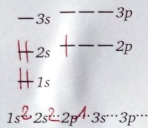
3 lithium Li



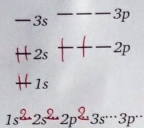
4 béryllium Be



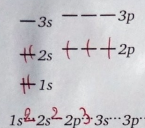
5 bore B



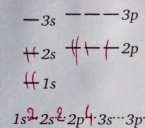
6 carbone C



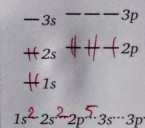
7 azote N



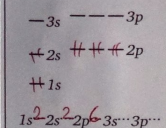
8 oxygène O



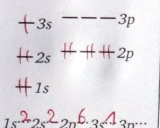
9 fluor F



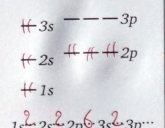
10 néon Ne



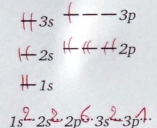
11 sodium Na



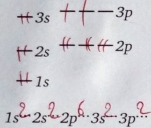
12 magnésium Mg



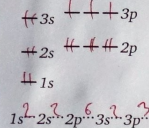
13 aluminium Al



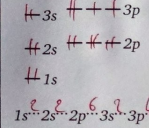
14 silicium Si



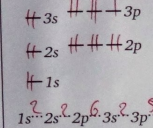
15 phosphore P



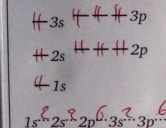
16 soufre S



17 chlore Cl



18 argon Ar

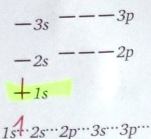


*Complétez le tableau avec*

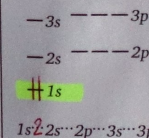
*- le numéro des couches*

*- le nombre d'électrons de valence*

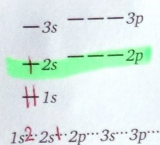
1 hydrogène H



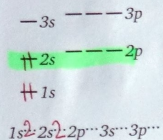
2 hélium He



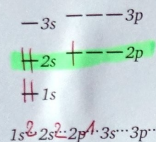
3 lithium Li



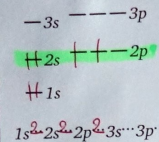
4 béryllium Be



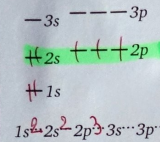
5 bore B



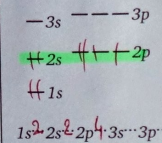
6 carbone C



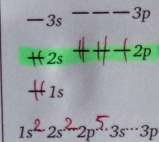
7 azote N



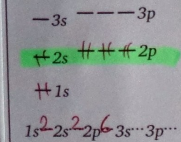
8 oxygène O



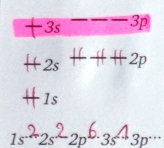
9 fluor F



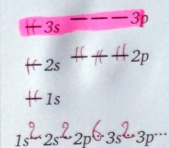
10 néon Ne



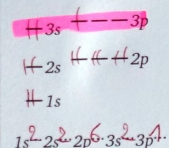
11 sodium Na



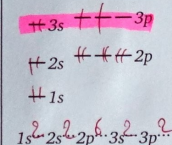
12 magnésium Mg



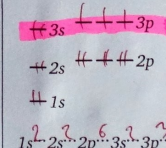
13 aluminium Al



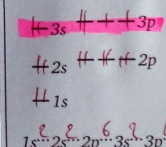
14 silicium Si



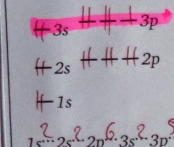
15 phosphore P



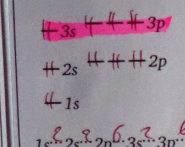
16 soufre S



17 chlore Cl



18 argon Ar



1

2

3

4

5

6

7

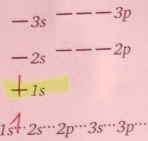
8

électrons de valence

*Complétez le tableau avec*

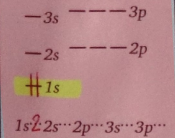
*- les blocs s et p*

1 hydrogène H



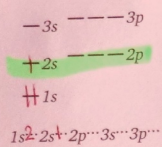
bloc s

2 hélium He

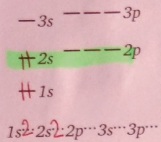


bloc p

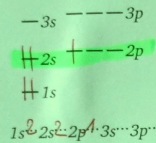
3 lithium Li



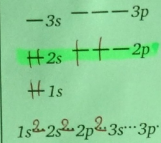
4 béryllium Be



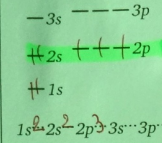
5 bore B



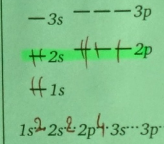
6 carbone C



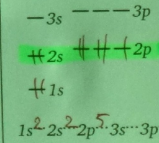
7 azote N



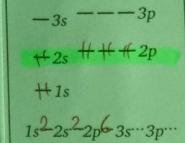
8 oxygène O



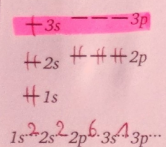
9 fluor F



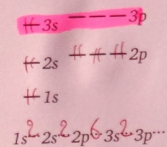
10 néon Ne



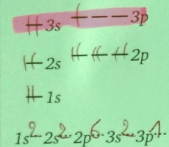
11 sodium Na



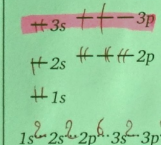
12 magnésium Mg



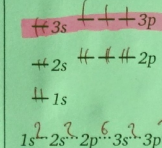
13 aluminium Al



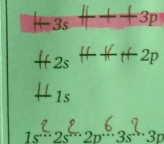
14 silicium Si



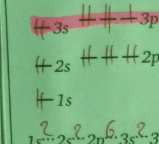
15 phosphore P



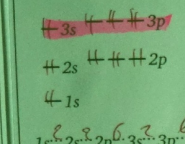
16 soufre S



17 chlore Cl



18 argon Ar



1

2

3

4

électrons

5

6

7

8

de valence